

Electrochemical capacitors for flexible energy storage applications

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INTRODUCTION

The rise of transparent and flexible electronics has driven interest in wearable devices [1], soft robotics [2], and e-skin [3]. Transparent energy storage devices are key to these applications, requiring high optical clarity and stable electrochemical performance [4]. Electrochemical capacitors are promising due to their ion-electron interaction-based operation. However, achieving high ionic conductivity and mechanical flexibility in transparent electrolytes remains a challenge. This work focuses on designing flexible transparent electrochemical capacitors (TECs) using hydrogel-based gel electrolytes and ITO electrodes.

EXPERIMENTAL

TECs were fabricated using ITO-coated PET substrates (>90% transparency, $0.0001 \Omega \cdot \text{cm}^{-1}$ sheet resistance) and NaCl/PVA, KCl/PVA, and LiCl/PVA gel polymer electrolytes (1:1 ratio). The gel electrolytes served as both ionic conductors and separators. Electrochemical properties were studied via CV and GCD, and optical transparency was evaluated using UV-Vis spectroscopy.

RESULTS AND DISCUSSION

LiCl/PVA and KCl/PVA gel electrolytes exhibited excellent optical transmittance (~98–100%) across the visible range, while NaCl/PVA showed slightly lower values (~95–99%). Among the samples, KCl/PVA demonstrated the highest ionic conductivity (98.8 mS/cm). Electrochemical impedance spectroscopy revealed solution resistance values of 12.17, 25.8, and $15.72 \Omega \cdot \text{cm}^2$ for NaCl, KCl, and LiCl-based gels, respectively. Cyclic voltammetry indicated diffusion-controlled charge storage in LiCl-based devices (slope = 0.44), and its optical bandgap was determined to be 5.37 eV.

CONCLUSIONS

Among the tested systems, the TEC employing the LiCl/PVA gel electrolyte demonstrated superior electrochemical and optical performance. It achieved a specific capacitance of 6.61 mF cm^{-2} at a current density of $10 \mu\text{A cm}^{-2}$ and a total charge contribution of 4.59 mC cm^{-2} , attributed predominantly to diffusion-controlled processes. The electrolyte exhibited 99% transparency at 550 nm and a direct optical bandgap of 5.34 eV, making it highly suitable for transparent energy storage applications.

[More details will be presented at the conference.]

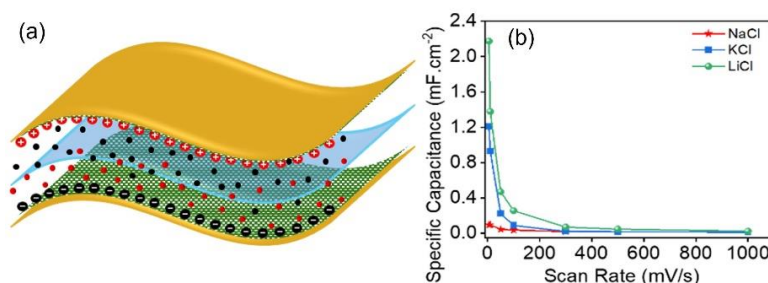


Fig. 1. (a) Schematic illustration of the TEC (b) Variation of specific capacitance with scan rate

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REFERENCES

1. J.H. Kim, et al., , Adv Funct Mater 31 (2021)
2. Z. Shen, et al., , Advanced Materials 33 (2021)
3. M. Zhang, et al., , Adv Funct Mater 33 (2023)
4. C. Kumar, et al., , Appl Phys Lett 124 (2024)